

## GLAUCIN B, A NEW BITTER LIMONOID FROM *EVODIA GLAUCA*

MUNEHIRO NAKATANI,\* HIDEKI TAKAO, TAKASHI IWASHITA,† HIDEO NAKOI† and TSUNAO HASE

Department of Chemistry, Faculty of Science, Kagoshima University, 1-21-35 Koorimoto, Kagoshima 890, Japan, †Suntory Institute for Bioorganic Research, Shimamoto-cho, Mishima-gun, Osaka 618, Japan

(Received 26 August 1987)

**Key Word Index**—*Evodia glauca*; Rutaceae; bitter limonoid, limonin, rutaevin, limonin diosphenol, glaucin A, 5 $\beta$ -H limonoid

**Abstract**—A new bitter limonoid, glaucin B, has been isolated from the root bark of *Evodia glauca* together with the known limonoids limonin, rutaevin, limonin diosphenol and glaucin A (12 $\alpha$ -hydroxyrutaevin). The structure of glaucin B determined to have the uncommon 5 $\beta$ -H configuration by spectral and chemical evidence.

### INTRODUCTION

In the course of our studies on the biologically active limonoids from Meliaceae and Rutaceae plants [1-5], we have isolated a new limonoid glaucin B along with the known limonoids, limonin, rutaevin, limonin diosphenol and glaucin A (12 $\alpha$ -hydroxyrutaevin), as bitter components of the root bark of *Evodia glauca* Miq., which is a large tree found in the southwest islands of Japan and resembles the Japanese Meliaceae *Melia azedarach* L. var. *Japonica* Makino. The structure and stereochemical assignments of glaucin B were made using COSY and NOESY spectroscopy, which indicated the uncommon 5 $\beta$ -H configuration of the new compound which is opposite to that of many known Rutaceae limonoids including limonin. The structure was related chemically to the known limonoids. Some biological activities of the limonoids isolated in this work were also examined.

### RESULTS AND DISCUSSION

The ether extract of the root bark (1 kg) contained many limonoids which were detected by the characteristic colour with Ehrlich's reagent on TLC. Fractionation by extensive silica gel column chromatography followed by the final HPLC purification on a normal phase column afforded a new limonoid glaucin B (1) along with limonin [6], rutaevin [7], limonin diosphenol (3) [8, 9] and glaucin A [5].

Glaucin B (1), mp 228-231° as powder from MeOH,  $[\alpha]_D^{25} + 29^\circ$  (MeOH,  $c$  0.001), was assigned the molecular formula  $C_{28}H_{32}O_{10}$  from FDMS ( $m/z$  529  $[M+H]^+$ ). Its UV, IR and CD spectra showed the presence of furan,  $\delta$ - and  $\gamma$ -lactone, ester and ketone groups. The  $^1H$  NMR spectrum of 1 showed signals of four tertiary methyl groups at  $\delta$  1.26 (8-Me), 1.28 (4 $\beta$ -Me), 1.32 (13-Me) and 1.37 (4 $\alpha$ -Me), one acetyl singlet at 2.21, a characteristic sharp epoxide proton singlet at 3.95 (H-15), an AB quartet ( $J$  = 15 Hz) at 4.07 (H-19a) and 4.31 (H-19b), and a singlet at 5.61 (H-17), along with the usual limonoid

furan signals at 7.44 ( $dd$ ,  $J$  = 1.5 and 1 Hz; H-21), 6.31 ( $dd$ ,  $J$  = 1.5 and 1 Hz; H-22) and 7.42 ( $t$ ,  $J$  = 1.5 Hz; H-23). The H-17 signal showed small allylic couplings with the H-21 and H-22 signals [2-5] and the H-19a signal showed a  $W$ -type long-range coupling with a signal at  $\delta$  4.25 (1H,  $br\ t$ ,  $J$  = 3 Hz; H-1), which revealed the H-1 to be in the  $\alpha$ -configuration [5]. Furthermore, the  $^1H$  NMR spectrum showed an acetoxy methine proton at  $\delta$  5.54 ( $d$ ,  $J$  = 11 Hz, H-6) coupled with a doublet methine proton at  $\delta$  2.34 ( $d$ ,  $J$  = 11 Hz; H-5). 2D-Homonuclear  $J$ -correlation (COSY) spectrum clarified the coupling protons on the linked carbons (C-1 and C-2, C-5 and C-6, and C-9, C-11 and C-12). These  $^1H$  NMR spectral features resembled those of limonin except for the presence of the signals due to an additional acetoxy group (see Experimental).

In the  $^1H$  NOE experiments on 1, irradiation of the 8-Me protons at  $\delta$  1.26 induced 6.9, 4.1, 2.2 and 1.3 % peak enhancements on the H-15 $\alpha$ , H-19a, H-19b and H-5 signals, respectively. The NOE on the H-5 suggested the  $\beta$ -configuration which is opposite to that of limonin, in the spectrum of which irradiation of the 8-Me protons at  $\delta$  1.07 showed the following NOEs: 5.5% (H-6 $\beta$ ), 13.5% (H-15), 2.5% (H-19a) and 6.7% (H-19b). The 5 $\beta$ -H configuration was also supported by 10% peak enhancement on the H-5 signal by irradiation of the 4 $\beta$ -Me protons at  $\delta$  1.28. On the other hand, irradiation of the 4 $\alpha$ -Me protons at  $\delta$  1.37 induced 4.9 and 10.2% peak enhancements on the H-1 $\alpha$  and H-6 signals, respectively, which clarified the  $\beta$ -configuration of the 6-acetoxy group. Therefore, compound 1 should be the 6 $\beta$ -acetoxy-5 $\beta$ -H analogue of limonin, which has the same A/B *cis*-ring junction with that of rutaevin and glaucin A. Irradiation of the 4 $\alpha$ -Me ( $\delta$  1.21) and 4 $\beta$ -Me ( $\delta$  1.44) protons in the spectrum of rutaevin induced 9.5 and 12.0% NOEs on the H-1 $\alpha$  and H-5 signals, respectively. Detailed NOE data on rutaevin and glaucin A are given in the Experimental section. The  $^1H$  NMR spectrum of 1 (Table 1) supports the structure and the cross-relaxation correlated 2D- $^1H$  NOE (NOESY) (4 $\alpha$ -Me/H-5, H-15, H-19a, H-19b and 13-Me/H-9, H-21, H-22) revealed the gross conformation (Fig. 1).

Hydrolysis of 1 with 0.1 N aq. sodium carbonate in MeOH followed by Jones oxidation, afforded an  $\alpha$ -diketone.

\* Author to whom correspondence should be addressed

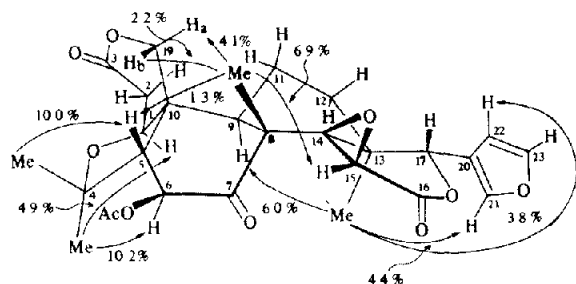


Fig. 1 Glaucin B, stereochemistry and NOEs between methyl and proton

tone **2**, mp 283–286°, CD(MeOH)  $\Delta\epsilon_{407} - 1.0$  ( $n-\pi^*$  of diketone), which was also derived from rutaevin by Jones oxidation (Scheme 1). Thus compound **1** was demonstrated to have the same stereochemistry as rutaevin. These reactions and the  $^1\text{H}$  NMR data led to the assignment of the structure of glaucin B as **1**.

When left in methanol for 1 day, the dione **2** was partly converted into a diosphenol **3** (limonin diosphenol) to give a mixture as shown from spectral data [ $^1\text{H}$  NMR  $\lambda_{\text{max}}^{\text{MeOH}}$  208 ( $\pi-\pi^*$  of furan), 277 nm ( $\pi-\pi^*$  of **3**); CD (MeOH)  $\Delta\epsilon_{211}$  ( $\pi-\pi^*$  of furan),  $\Delta\epsilon_{272}$  ( $\pi-\pi^*$  of **3**),  $\Delta\epsilon_{319}$  ( $n-\pi^*$  of **2**) (Fig. 2). On the other hand, oxidation of the hydrolysis product from **1** with pyridinium chlorochromate afforded the diosphenol **3**, which was also obtained from the dione **2** by treatment with alkali. The formation of compound **3** has been reported from limonin and rutaevin [7, 8].

Although there have been many Rutaceous limonoids reported, the  $5\beta$ -H compounds are rare [5, 10]. Among

the compounds isolated in this work, limonin and limonin diosphenol (**3**) were active as an insect antifeedant against the larvae of *Spodoptera litura* Fab., but glaucin B (**1**) showed no activity. On the other hand, in antimicrobial tests against bacteria, yeasts and fungi rutaevin showed weak antibacterial activity against *Bacillus subtilis*.

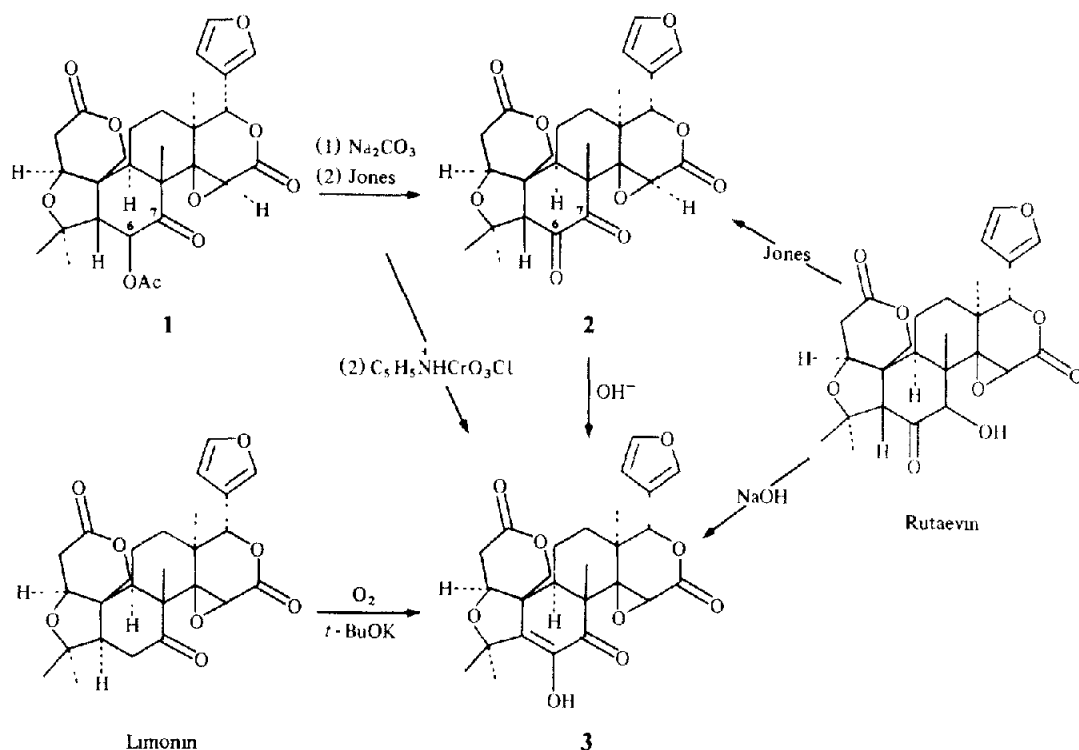
## EXPERIMENTAL

Mps uncorr. UV spectra were measured in MeOH and  $^1\text{H}$  NMR spectra in  $\text{CDCl}_3$  with TMS as int. standard were measured at 360 MHz.

**Plant material.** Roots of the plant were collected at the botanical garden of Kagoshima University in August 1983 and identified by Dr S. Sako (Kagoshima University).

**Extraction and isolation.** Fresh root bark (10 kg) was extracted for 1 week with MeOH ( $3 \times 3\text{ l}$ ) at room temp. After concn to 200 ml,  $\text{H}_2\text{O}$  (300 ml) was added to afford 5.2 g of a ppt. which was applied onto a silica gel column. The column was eluted with a  $\text{CH}_2\text{Cl}_2$ -MeOH gradient to yield a complex limonoid mixture and glaucin A (13 mg). The mixture was separated by repeated passage through a HPLC, semiprep. silica gel column, using 0.4–1.6% MeOH- $\text{CH}_2\text{Cl}_2$  as the solvent to give **1** (26 mg), limonin (320 mg), rutaevin (26 mg) and limonin diosphenol (**3**, 21 mg).

**Glaucin B (1).** Powder from MeOH, mp 228–231°,  $[\alpha]_D^{25} + 29^\circ$  (MeOH,  $c$  0.001), UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ) 209 (3.72), 279 (2.43), IR  $\nu_{\text{max}}^{\text{NaOH}}$   $\text{cm}^{-1}$  1760 ( $\gamma$ -lactone), 1735 ( $\delta$ -lactone and ester), 1700 (ketone), CD (MeOH) nm  $\Delta\epsilon_{211} - 4.0$  ( $\pi-\pi^*$  of furan),  $\Delta\epsilon_{234} - 4.7$  ( $n-\pi^*$  of lactone and ester),  $\Delta\epsilon_{296} + 5.9$ ,  $\Delta\epsilon_{305} + 5.6$ ,  $\Delta\epsilon_{315} + 2.3$  ( $n-\pi^*$  of ketone), FDMS  $m/z$  529  $[\text{M} + 1]^+$ , 469, 431, 265, 183, 129,  $^1\text{H}$  NMR see Table I.



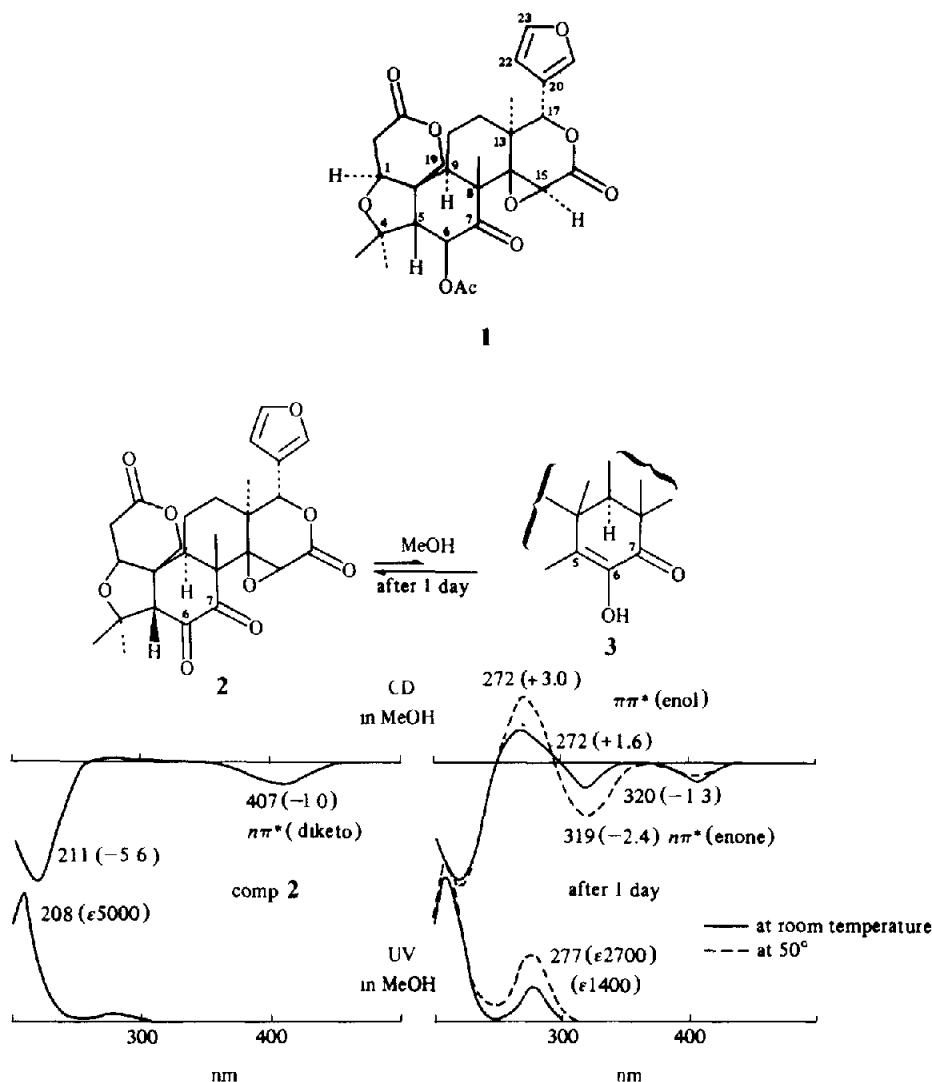


Fig. 2

Table 1  $^1\text{H}$  NMR data of Glaucin B (1) ( $\text{CDCl}_3$ , 360 MHz)

H	$\delta$	Mult	$J$ (Hz)	Coupled to	H	$\delta$	Mult	$J$ (Hz)	Coupled to
1	4.25	<i>brt</i>	3	2a, 2b, (19a)*	17	5.61	<i>brs</i>		(21), (22)
2a	2.64	<i>dd</i>	3, 16	1, 2b	18	1.32	<i>s</i>		
2b	2.89	<i>dd</i>	3, 16	1, 2a	19a	4.07	<i>brd</i>	15	19b, (1)
5	2.34	<i>d</i>	11	6	19b	4.31	<i>d</i>	15	19a
6	5.54	<i>d</i>	11	5	21	7.44	<i>brdd</i>	1, 1.5	22, 23, (17)
9	3.19	<i>dd</i>	5, 13	11 $\alpha$ , 11 $\beta$	22	6.31	<i>brdd</i>	1, 1.5	21, 23, (17)
11 $\alpha$	1.79	<i>m</i>		9, 11 $\beta$ , 12 $\alpha$ , 12 $\beta$	23	7.42	<i>t</i>	1.5	21, 22
11 $\beta$	1.93	<i>m</i>		9, 11 $\alpha$ , 12 $\alpha$ , 12 $\beta$	28	1.37	<i>s</i>		
12 $\alpha$	1.63	<i>m</i>		11 $\alpha$ , 11 $\beta$ , 12 $\beta$	29	1.28	<i>s</i>		
12 $\beta$	1.82	<i>m</i>		11 $\alpha$ , 11 $\beta$ , 12 $\alpha$	30	1.26	<i>s</i>		
15	3.95	<i>s</i>			OAc	2.21	<i>s</i>		

\*Distinguished couplings are listed and numerals in parenthesis denote small coupling based on COSY.

**Limonin.** Prisms from MeOH; mp 289–293°;  $[\alpha]_D^{24} -136^\circ$  ( $\text{Me}_2\text{CO}$ ,  $c$  0.004); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm(log  $\epsilon$ ). 208(3.86); IR  $\nu_{\text{max}}^{\text{Nujol}}$   $\text{cm}^{-1}$  1760, 1705, 1503, 878; CD(MeOH) nm  $\Delta\epsilon_{230} -4.9$ ,  $\Delta\epsilon_{293} -2.3$ ; EIMS  $m/z$ : 470  $[\text{M}]^+$ , 455, 426, 413, 347, 135, 108, 94,  $^1\text{H}$  NMR:  $\delta$  1.07 (3H, *s*, 8-Me), 1.18 (3H, *s*, 4 $\beta$ -Me), 1.18 (3H, *s*, 13-Me), 1.30 (3H, *s*, 4 $\alpha$ -Me), 1.52 (1H, *m*, H-12 $\alpha$ ), 1.78 (1H, *m*, H-12 $\beta$ ), 1.83 (1H, *m*, H-11 $\alpha$ ), 1.90 (1H, *m*, H-11 $\beta$ ), 2.23 (1H, *dd*,  $J = 16$  and 3 Hz, H-5), 2.47 (1H, *dd*,  $J = 14.5$  and 3.5 Hz, H-6 $\alpha$ ), 2.55 (1H, *dd*,  $J = 12$

and 2.5 Hz, H-9), 2.68 (1H, *dd*,  $J = 17$  and 2 Hz, H<sub>a</sub>-2), 2.86 (1H, *dd*,  $J = 15.5$  and 14.5 Hz, H-6 $\beta$ ), 2.98 (1H, *dd*,  $J = 17$  and 4 Hz, H<sub>b</sub>-2), 4.03 (1H, *m*, H-1), 4.03 (1H, *s*, H-15), 4.47 (1H, *d*,  $J = 13$ , H<sub>a</sub>-19), 4.77 (1H, *br d*,  $J = 13$ , H<sub>b</sub>-19), 5.47 (1H, *br s*, H-17), 6.34 (1H, *ddd*,  $J = 2, 1$  and 1 Hz, H-22), 7.40 (1H, *dd*,  $J = 2$  and 1.5 Hz, H-23), 7.42 (1H, *dd*,  $J = 1.5$  and 1 Hz, H-21). NOEs (no indication in parentheses based on 2D-NOE) H-9/H-1, H-5, 13-Me, 13-Me/H-9 (5.8%), H-21 (4.3%), H-22 (5%), 4 $\alpha$ -Me/H-5 (6.9%), H-1 (2.5%), 4 $\beta$ -Me/H-19b, 8-Me/H-6 $\beta$  (5.5%), H-15 (13.5%), H<sub>a</sub>-19 (2.5%), H<sub>b</sub>-19 (6.7%).

**Rutaevin** Needles from MeOH, mp > 300°,  $[\alpha]_D^{24} - 146^\circ$  (MeOH,  $c$  0.0005), UV  $\lambda_{max}$  nm (log  $\epsilon$ ) 208 (3.97), IR  $\nu_{max}^{Nujol}$   $cm^{-1}$  3420, 1770, 1740, 1710, 1504, 880, CD (MeOH) nm  $\Delta\epsilon_{230} - 4.1$ ,  $\Delta\epsilon_{295} - 1.0$ , FDMS  $m/z$  487  $[M + 1]^+$ , 410, 369, 327, 244, 190, 127, 85,  $^1H$  NMR  $\delta$  0.65 (3H, *s*, 8-Me), 1.21 (3H, *s*, 4 $\alpha$ -Me), 1.36 (3H, *s*, 13-Me), 1.44 (3H, *s*, 4 $\beta$ -Me), 1.6 (1H, *m*, H-12 $\alpha$ ), 1.73 (1H, *br dd*,  $J = 13$  and 8 Hz, H-11 $\alpha$ ), 1.85 (1H, *m*, H-11 $\beta$ ), 1.95 (1H, *m*, H-12 $\beta$ ), 2.64 (1H, *dd*,  $J = 15$  and 3 Hz, H<sub>a</sub>-2), 2.79 (1H, *br d*,  $J = 12$  Hz, H-9), 2.92 (1H, *dd*,  $J = 15$  and 3 Hz, H<sub>b</sub>-2), 3.12 (1H, *s*, H-5), 3.89 (1H, *d*,  $J = 3$  Hz, -OH), 4.15 (1H, *d*,  $J = 12.5$  Hz, H<sub>a</sub>-19), 4.16 (1H, *s*, H-15), 4.28 (1H, *br d*,  $J = 12.5$  Hz, H<sub>b</sub>-19), 4.35 (1H, *br d*,  $J = 3$  Hz, H-7), 4.37 (1H, *br t*,  $J = 3$  Hz, H-1), 5.48 (1H, *br s*, H = 17), 6.35 (1H, *m*, H-22), 7.43 (1H, *m*, H-23), 7.44 (1H, *m*, H-21). NOEs 13-Me/H-7 (5.8%), H-9 (6.1%), H-21 (4.2%), H-22 (6.1%), 4 $\alpha$ -Me/H-1 (9.5%), 4 $\beta$ -Me/H-5 (12.0%), 8-Me/H-15 (18.4%), H<sub>a</sub>-19 (3.6%), H<sub>b</sub>-19 (11.3%).

**Limonin diosphenol (3)** Needles from Me<sub>2</sub>CO, mp 273–278°,  $[\alpha]_D^{24} - 193^\circ$  (Me<sub>2</sub>CO,  $c$  0.002), UV  $\lambda_{max}$  nm (log  $\epsilon$ ) 207 (3.90), 278 (3.90), IR  $\nu_{max}^{Nujol}$   $cm^{-1}$  3450, 1740, 1690, 1660, 880, CD (MeOH) nm  $\Delta\epsilon_{208} - 4.2$ ,  $\Delta\epsilon_{275} + 10.7$ ,  $\Delta\epsilon_{320} - 8.5$ ,  $^1H$  NMR  $\delta$  1.04 (3H, *s*, 8-Me), 1.16 (3H, *s*, 4 $\beta$ -Me), 1.49 (3H, *s*, 13-Me), 1.54 (3H, *s*, 4 $\alpha$ -Me), 2.64 (1H, *dd*,  $J = 13$  and 3 Hz, H-9), 2.84 (1H, *dd*,  $J = 17$  and 5 Hz, H<sub>a</sub>-2), 2.91 (1H, *dd*,  $J = 17$  and 3 Hz, H<sub>b</sub>-2), 4.07 (1H, *br t*,  $J = 3$  Hz, H-1), 4.12 (1H, *s*, H-15), 4.62 (1H, *d*,  $J = 13$  Hz, H<sub>a</sub>-19), 4.66 (1H, *d*,  $J = 13$  Hz, H<sub>b</sub>-19), 5.43 (1H, *s*, H-17), 6.23 (1H, *s*, -OH), 6.38 (1H, *m*, H-22), 7.40 (1H, *m*, H-23), 7.41 (1H, *m*, H-21).

**Glaucin A** Prisms from acetonitrile, mp > 310°,  $[\alpha]_D^{15} - 150^\circ$  (MeCN,  $c$  0.0005), UV  $\lambda_{max}$  nm (log  $\epsilon$ ) 209 (3.79), IR  $\nu_{max}^{Nujol}$   $cm^{-1}$  3450, 1765, 1740, 1700, 1503, 1110, 1050, 1010, 875, CD (MeOH)  $\Delta\epsilon_{228} - 5.3$ ,  $\Delta\epsilon_{297} - 1.7$ , SIMS  $m/z$  503  $[M + H]^+$ , 645, 595, 553, 461, 277, 185, 93, 75, 57,  $^1H$  NMR  $\delta$  0.61 (3H, *s*, 8-Me), 1.23 (3H, *s*, 4 $\alpha$ -Me), 1.35 (3H, *s*, 13-Me), 1.44 (3H, *s*, 4 $\beta$ -Me), 2.63 (1H, *dd*,  $J = 15$  and 3 Hz, H<sub>a</sub>-2), 2.91 (1H, *dd*,  $J = 15$  and 3 Hz, H<sub>b</sub>-2), 2.98 (1H, *br d*,  $J = 12$  Hz, H-9), 3.13 (1H, *s*, H-5), 3.87 (1H, *d*,  $J = 3$  Hz, -OH), 4.04 (1H, *m*, H-12 $\beta$ ), 4.10 (1H, *d*,  $J = 12$  Hz, H<sub>a</sub>-19), 4.15 (1H, *s*, H-15), 4.28 (1H, *d*,  $J = 12$  Hz, H<sub>b</sub>-19), 4.34 (1H, *m*, H-7), 4.37 (1H, *d*,  $J = 7$  Hz, -OH), 4.45 (1H, *m*, H-1), 5.51 (1H, *s*, H-17), 6.46 (1H, *m*, H-22), 7.29 (2H, *m*, H-21 and H-23) (Found C, 62.12, H, 6.04%. Calcd for C<sub>26</sub>H<sub>30</sub>O<sub>10</sub> C, 62.14, H, 6.02%).

**$\alpha$ -Diketone 2 from glaucin B (1)** To a MeOH soln (10 ml) of 1 (18 mg), 0.1 N Na<sub>2</sub>CO<sub>3</sub> soln (3 ml) was added and the mixture stirred at room temp for 2 hr. Work-up as usual gave 15 mg of crude product. The product (8 mg) was stirred with Jones reagent in Me<sub>2</sub>CO at room temp for 1 hr. Work-up as usual gave 6 mg of crude product, which was purified by column chromatography on silica gel and HPLC using a normal phase column to give 2 (3 mg) as pale yellow prisms, mp 282–286° (dec.). UV  $\lambda_{max}$  nm (log  $\epsilon$ ) 208 (3.70), CD (MeOH) nm  $\Delta\epsilon_{211} - 5.6$  ( $\pi - \pi^*$  of furan),  $\Delta\epsilon_{407} - 1.0$  ( $n - \pi^*$  of diketo) (Found C, 64.51, H, 5.81%. Calcd for C<sub>26</sub>H<sub>28</sub>O<sub>9</sub> C, 64.45, H, 5.83%). To a CH<sub>2</sub>Cl<sub>2</sub> soln of the hydrolysed product (7 mg) of 1, pyridinium chlorochromate (30 mg) was added and stirred at room temp. After 1 hr, MeOH (1 ml) was added and the mixture stirred for an additional 1 hr. The crude product was chromatographed and purified by HPLC to give 4 mg of limonin diosphenol (3).

**$\alpha$ -Diketone 2 from rutaevin [7]** Rutaevin (7 mg) was treated with Jones reagent in Me<sub>2</sub>CO at room temp in a similar manner described above to give 2 (4 mg) as pale yellow prisms, mp 283–286° (dec.).

**Limonin diosphenol (3) from limonin [8]** Limonin (100 mg) suspended in *t*-butyl alcohol: *N* potassium *t*-butoxide (20 ml) was shaken with oxygen in a hydrogenation apparatus for 2 hr. After addition of H<sub>2</sub>O, the soln was acidified with 6 N HCl soln and extracted with CHCl<sub>3</sub>. The CHCl<sub>3</sub> soln was washed with NaHCO<sub>3</sub> soln and H<sub>2</sub>O, and then shaken with 4N NaOH soln. Acidification of this extract with HCl soln gave a ppt which was extracted with CHCl<sub>3</sub>. The CHCl<sub>3</sub> soln was washed with H<sub>2</sub>O, and the solvent was removed. Recrystallization from Me<sub>2</sub>CO gave 3 (43 mg) as needles, mp 273–278°.

**Limonin diosphenol (3) from rutaevin** Pyridinium chlorochromate (80 mg) was added to a stirred soln of rutaevin (44 mg) in CH<sub>2</sub>Cl<sub>2</sub>. After stirring over night at room temperature, MeOH was added and the mixture stirred for an additional 1 hr, and then the soln was evaporated *in vacuo* to yield a brown residue which was chromatographed over silica gel to give 3 (30 mg).

**Biological activities** (i) *Insect antifeedant activity* five limonoids, limonin, rutaevin, limonin diosphenol, glaucin A and glaucin B, were tested against the larvae of the pest insect *Spodoptera litura* Fab. with the leaf disk method [11]. Limonin and limonin diosphenol (3) showed the activity only at 1000 ppm concn. (ii) *Antimicrobial activity* Effects of the compounds on growth of microorganisms were tested by the broth dilution method by Dr M. Taniguchi (Osaka City University). Among the compounds rutaevin was active against a bacterium *Bacillus subtilis* at 100  $\mu$ g/ml concn. Test organisms bacterium, *S. aureus*, *B. subtilis*, *E. coli* and *Ps. aeruginosa* fungus, *Mucor mucedo*, *Rhizopus*, *Asp. niger* and *P. crustosum* yeast, *S. cerevisiae*, *C. utilis*, *Schiz. pombe* and *H. anomala*.

**Acknowledgements**—We are grateful to Dr S. Sako (Kagoshima University) for the identification of the plant and to Mr M. Tahara (Sankai Chemical Co., Ltd.) for the supply of the insect. Our thanks are also due to Dr M. Taniguchi (Osaka City University) for the antimicrobial test and Dr K. Shibata (Osaka City University) for the elemental analysis.

## REFERENCES

- 1 Nakatani, M., James, J. C. and Nakanishi, K. (1981) *J. Am. Chem. Soc.* **103**, 1228.
- 2 Nakatani, M., Okamoto, M., Iwashita, T., Mizukawa, K., Naoki, H. and Hase, T. (1984) *Heterocycles* **22**, 2335.
- 3 Nakatani, M., Iwashita, T., Naoki, H. and Hase, T. (1985) *Phytochemistry* **24**, 195.
- 4 Nakatani, M., Iwashita, T., Mizukawa, K. and Hase, T. (1987) *Heterocycles* **26**, 43.
- 5 Nakatani, M., Takao, H., Iwashita, T., Naoki, H. and Hase, T. (1987) *Bull. Chem. Soc. Jpn* **60**, 2503.
- 6 Arigoni, D., Barton, D. H. R., Corey, E. J., Jeger, O., Cogholi, L., Ferrini, S. G., Glazier, E. R., Melera, A., Pradhan, S. K., Schaffner, K., Sternhell, S., Templeton, J. E. and Tobinaga, S. (1960) *Experimentia* **16**, 41.
- 7 Dreyer, D. L. (1967) *J. Org. Chem.* **32**, 3442.
- 8 Barton, D. H. R., Pradhan, S. K., Sternhell, S. and Templeton, J. F. (1961) *J. Chem. Soc.* 255.
- 9 Hirose, Y. (1963) *Chem. Pharm. Bull. Tokyo* **11**, 535.
- 10 Bennett, R. D. and Hasegawa, S. (1981) *Tetrahedron* **37**, 17.
- 11 Kubo, I., Taniguchi, M., Chapya, A. and Tsujimoto, K. (1980) *Planta Med.* **185**.